

## The prototype of a chemical computer detects a sphere

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A chemical excitation is being transmitted from the droplet in the lower left corner to two adjacent droplets. Images have been taken through a Hirox KH-8700 microscope. Credit: IPC PAS, Hirox

Researchers in Warsaw have developed a system that uses chemistry to make computations. After an appropriate teaching procedure, even a relatively simple chemical system can perform non-trivial operations. In their most recent computer simulations, the researchers have shown that correctly programmed chemical matrices of oscillating droplets can recognize the shape of a sphere with great accuracy.

Modern computers exploit physical phenomena related to the movement of electric charges to make calculations. Information can, however, be processed in many ways. Efforts have been underway worldwide to use chemical signals for this purpose. Currently, such chemical systems perform only the simplest logic operations. Now, researchers from the Institute of Physical Chemistry of the Polish Academy of Sciences (IPC PAS) have demonstrated that even uncomplicated and easy-to-produce collections of droplets, in which oscillating chemical reactions proceed, can process information in a useful way, e.g. recognizing the shape of a specified three-dimensional object with great accuracy or correctly classifying cancer cells as benign or malignant.

"A lot of work being carried out in laboratories focuses on building chemical equivalents of standard logic gates. We took a different approach to the problem," says Dr. Konrad Gizynski. "We investigate systems of a few dozen drops in which chemical signals propagate, and treat each one as a whole, as a kind of neuronal network. It turns out that such networks, even very simple ones, after a short teaching procedure,



manage well with fairly sophisticated problems. For instance, our newest system has ability to recognize the shape of a <u>sphere</u> in a set of x, y, z spatial coordinates."

The systems being studied at the IPC PAS work thanks to the Belousov-Zhabotinsky reaction proceeding in individual drops. This reaction is oscillatory: After the completion of one oscillation cycle, the reagents necessary to begin the next cycle are regenerated in the solution. A droplet is a batch reactor. Before reagents are depleted, a droplet has usually performed from a few dozen to a few hundred oscillations. The time evolution of a droplet is easy to observe, since its catalyst, ferroin, changes colour during the cycle. In a thin layer of solution, the effect is spectacular: Colourful strips—chemical fronts—traveling in all directions appear in the liquid. Fronts can also be seen in the droplets, but in practice, the phase of the cycle is indicated just by the colour of the droplet: when the cycle begins, the droplet rapidly turns blue (excites), after which it gradually returns to its initial red state.

"Our systems basically work by mutual communication between droplets. When the droplets are in contact, the chemical excitation can be transmitted from droplet to droplet. In other words, one droplet can trigger the reaction in the next. It is also important that an excited droplet cannot be immediately excited once again. Speaking somewhat colloquially, before the next excitation it has to 'have a rest' in order to return to its original state," explains Dr. Gizynski.

Information has to be introduced into the system, and after processing, it has to be read out. It is also important to have the ability to modify the way the system processes information in a controlled manner. To meet these challenges, researchers from the IPC PAS used light and an additional (apart from ferroin) catalyst, ruthenium. The Belousov-Zhabotinsky reaction catalyzed by ruthenium has an important feature: It is inhibited by blue light, which means that with intensive illumination,



the droplets cease to oscillate. By changing the illumination of a specific droplet, it is possible to decide whether it is to be involved in the processing of information or not, and to excite it according to any selected pattern. In the case of droplets used to input information, a longer illumination time means a greater value of the spatial coordinate to be entered.

"In practice, the illumination of droplets can be controlled individually by attaching a separate optical fiber to each droplet," says Dr. Gizynski, and stresses that all the simulation parameters—transitions between droplet states, the propagation of chemical fronts between droplets, stopping and resuming the reactions by the light brought in by the fiber as well as the production of networks of droplets—were selected on the basis of previously performed experiments.

In their simulations, researchers from Warsaw looked for methods of detecting the shape of a sphere, considering networks composed of 2x2, 3x3, 4x4 and 5x5 droplets. To map the actual rates of the chemical reactions, it was assumed that a droplet spends one second in the excited state and next returns to a state in which it can be activated again within 10 seconds.

Before starting the teaching procedure, the researchers also had to create the right input with a description of the sphere. For this purpose, they generated a random set of points which were assigned a value of 1 when the given point belonged to the sphere or 0 when the point lay beyond it. A database prepared in this manner became the basis of the teaching process, in which each of the components of the point (x, y, z)determined the time of illumination of different input droplets.

To teach the system of droplets to detect the shape of a sphere, researchers from the IPC PAS used evolutionary algorithms. The teaching procedure began with the random generation of 30 patterns of



illumination of the system, in which some <u>droplets</u> were treated as the input, while the others remained inactive for the time which was the subject of the optimization. After the droplet system had processed the entire database, the droplet in which time evolution was best correlated with the expected result was treated as the output. From the first generation of systems thus obtained, a few of the best were selected, multiplied by introducing small changes ('mutations') in the manners of illumination, and the teaching cycle started again from the beginning, continuing for 500 generations.

"We achieved the best results for the 4x4 droplet network. It showed the greatest accuracy in detecting the shape of a sphere, at a level of 85 percent. In addition, it acquired this ability the most rapidly, in just 150 generations. The 5x5 system could perhaps be better, but in order to check this, the teaching process should be extended to more than 500 generations," says Dr. Gizynski.

Droplet systems do not interpret the incoming data, they just look for correlations ('shapes') among them similar to the one they have been taught to find. Instead of the spatial coordinates of points with the shape of a sphere, they can be fed with data with a different meaning, e.g., associated with the different characteristics of tumour cells. The system would then be able to seek the 'shape' of the data corresponding to, for example, benign or malignant tumours.

"Indeed, in one of our recent publications, carried out in collaboration with colleagues from the University of Jena, we showed that the 5x5 droplet system could classify cancer cells from a machine learning dataset, the Wisconsin Breast Cancer Dataset, with a precision of up to 97 percent. Better results are achieved using conventional computers, but these also include classifiers acting less effectively. Thus, the chemical processing of information, although still heavily flawed, is beginning to offer more and more interesting and more useful possibilities,"



concludes Dr. Gizynski.

**More information:** Konrad Giżyński et al, A Chemical System that Recognizes the Shape of a Sphere, *Computational Methods in Science and Technology* (2016). DOI: 10.12921/cmst.2016.0000057

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