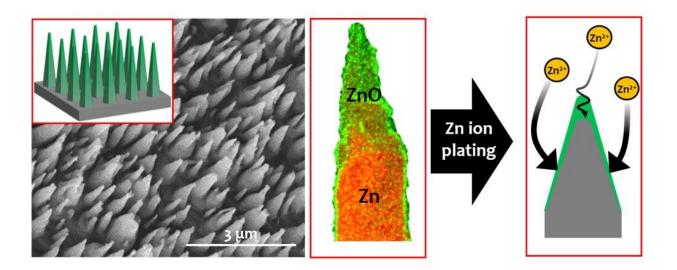


Development of next-generation zinc ion battery without the risk of explosion or fire

September 2 2020



Dendrite-free "functionalized ZnO layer" coated on a Zn hexagonal pyramid core. Credit: Korea Institue of Science and Technology (KIST)

A research team led by Dr. Joong-Kee Lee of the Center for Energy Storage Research has developed a next-generation secondary battery that uses zinc metal as an electrode without any risk of explosion or fire. This battery is safe enough to be worn on the body and can be manufactured as a fiber shape, which means it may potentially be applied as a power source for wearable devices in the future.



Recently, the demand for safe batteries has been rising dramatically, mainly due to fires occurring in electronic devices using <u>lithium-ion</u> <u>batteries</u>. Highly flammable electrolytes are the main cause of such fires, but since Zn-ion secondary batteries use water-based electrolytes, there is no risk of explosion. They are thus considered one of the more promising candidates to replace Li-ion batteries.

However, <u>zinc</u> anodes, which are the core material of existing Zn-ion batteries, present an inherent problem in that they undergo continuous corrosion in water-based electrolytes. Not only that, when zinc ions are stored on a <u>metal surface</u>, they accumulate as crystals in the form of branches (dendrites) and trigger a short circuit between electrodes, resulting in a sharp decline in efficiency. Studies have been carried out to come up with a solution to this problem through the means of a zinc <u>metal</u> complex, <u>surface coating</u>, and shape change, for example, but there have been major limitations in relation to processing cost and time.

The team headed by Dr. Lee from KIST developed a periodic anodizing method, which involves repeatedly permitting and blocking a flow of current on the surface of the metal electrode, thereby successfully controlling the surface coating morphologies and shape pattern array of the zinc oxide film simultaneously.

Using this method, the KIST research team inhibited the generation of dendrites during the electrochemical reaction by forming a functionalized shape in which hexagonal pyramids were arranged on the surface of zinc metal. According to the periodic anodizing method, the zinc oxide covering the upper part of the hexagonal pyramid is thick, whereas it is thin on the sides. The variation in thickness induces the zinc metal to accumulate on the side with a relatively thinner layer of zinc oxide. Dendrites are a problem as they accumulate vertically on the metal surface, but the newly developed technology in question induces the zinc metal film to grow in a horizontal direction on the electrode



surface, and it was able to effectively suppress the generation of dendrite. As for the zinc oxide film that formed on the <u>surface</u>, direct contact with electrolytes was blocked, thereby preventing corrosion and side reaction at the same time.

The Zn-ion secondary <u>battery</u> developed through this study maintained nearly 100% of its capacity over 1,000 cycles, even though it was repeatedly charged and discharged under <u>extreme conditions</u> (9,000 mA/g, fully charged and discharged for about two minutes each), attributed to its structural and electrochemical stability.

Based on such stability, the KIST researchers made a Zn-ion secondary battery in the form of flexible fibers. In addition to being bent can be bent easily, it can be used incorporated into clothing, a bag if it is made into fabric.

Dr. Lee, a senior researcher at KIST, said, "The high-performance Znion secondary battery developed in this study does not present any potential risks associated with Li-ion batteries coming into contact with the human body. At the same time, we expected it to garner attention as a next-generation secondary battery that is safe for the human body and doesn't present any risks of explosion or fire, along with its excellent electrochemical performance that is comparable to the existing commercial batteries in terms of battery capacity. It appears that based on excellent stability, improved electrochemical performance, and simple processes, it will be possible to make the manufacturing process practical for real-life applications."

More information: Ji Young Kim et al, Functionalized Zn@ZnO Hexagonal Pyramid Array for Dendrite-Free and Ultrastable Zinc Metal Anodes, *Advanced Functional Materials* (2020). DOI: <u>10.1002/adfm.202004210</u>



Provided by National Research Council of Science & Technology

Citation: Development of next-generation zinc ion battery without the risk of explosion or fire (2020, September 2) retrieved 24 April 2024 from <u>https://techxplore.com/news/2020-09-next-generation-zinc-ion-battery-explosion.html</u>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.