

Hybrid redox-flow battery with a long cycle life

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Redox-flow batteries store electrical energy in chemical compounds that are dissolved in an electrolyte. They are a particularly promising alternative to lithium-ion batteries as stationary energy storage. A team



headed by Prof. Dr. Ingo Krossing from the Institute of Inorganic and Analytical Chemistry at the University of Freiburg has succeeded in developing a non-aqueous All-Manganese Flow battery (All-MFB) that uses sustainable manganese as its active material and has a long cycle life. The researchers present the results of their work in the latest edition of *Advanced Energy Materials*.

Active materials are chemical substances that are required to store energy in batteries. The Freiburg scientists have now replaced the previous active material, the element vanadium, with the much more abundant element manganese. Krossing and his team have adopted a new approach to applying the sustainable manganese in the battery: Until now coupling the deposition of manganese in its elemental form with the oxidation of manganese in the oxidation state +II to manganese +III had not been used to store electrochemical energy. The newly-developed battery achieves an <u>energy density</u> roughly twice as high compared to the standard redox-flow battery with vanadium.

"With the electrolytes presented in our publication, energy densities of up to 74 Wh L–1 are possible," explains Krossing. "This was already far better in the first try than the energy density of the vanadium redox-flow battery which has been researched since 1978." Further optimization of the battery is necessary, according to the Freiburg chemist, "but this system describes a new and highly promising design for sustainable stationary energy storage."

More information: Maximilian Schmucker et al, Investigations toward a Non-Aqueous Hybrid Redox-Flow Battery with a Manganese-Based Anolyte and Catholyte, *Advanced Energy Materials* (2021). DOI: 10.1002/aenm.202101261



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