

Major advance in fabrication of low-cost solar cells also locks up greenhouse gases

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Perovskite solar cells have progressed in recent years with rapid increases in power conversion efficiency (from 3% in 2006 to 25.5% today), making them more competitive with silicon-based photovoltaic

cells. However, a number of challenges remain before they can become a competitive commercial technology.

Now a team at the NYU Tandon School of Engineering has developed a process to solve one of them, a bottleneck in a critical step involving p-type [doping](#) of organic hole-transporting materials within the photovoltaic [cells](#). The research, "CO₂ doping of organic interlayers for [perovskite solar cells](#)," appears in *Nature*.

Currently, the p-doping process, achieved by the ingress and diffusion of oxygen into the hole transporting layer, is time intensive (several hours to a day), making commercial mass production of [perovskite](#) solar cells impractical.

The Tandon team, led by André D. Taylor, an associate professor, and Jaemin Kong, a post-doctoral associate, along with Miguel Modestino, assistant professor—all in the Department of Chemical and Biomolecular Engineering—discovered a method of vastly increasing the speed of this key step through the use of carbon dioxide (CO₂) instead of oxygen.

In perovskite solar cells, doped organic semiconductors are normally required as charge-extraction interlayers situated between the photoactive perovskite layer and the electrodes. The conventional means of doping these interlayers involves the addition of lithium bis(trifluoromethane)sulfonimide (LiTFSI), a lithium salt, to spiro-OMeTAD, a π -conjugated organic semiconductor widely used for a hole-transporting material in perovskite solar cells. The doping process is then initiated by exposing spiro-OMeTAD:LiTFSI blend films to air and light.

Not only is this method time-consuming, it largely depends on ambient conditions. By contrast, Taylor and his team reported a fast and

reproducible doping method that involves bubbling a spiro-OMeTAD:LiTFSI solution with CO₂ under ultraviolet light. They found that their process rapidly enhanced electrical conductivity of the interlayer by 100 times compared to that of a pristine blend film, which is also approximately 10 times higher than that obtained from an oxygen bubbling process. The CO₂ treated film also resulted in stable, high-efficiency perovskite solar cells without any post-treatments.

"Besides shortening the device fabrication and processing time, application of the pre-doped spiro-OMeTAD in perovskite solar cells makes the cells much more stable," explained Kong, the lead author. "That's partly because most of the detrimental lithium ions in the spiro-OMeTAD:LiTFSI solution were stabilized as lithium carbonates during the CO₂ bubbling process."

He added that the lithium carbonates end up being filtered out when the investigators spincoat the pre-doped solution onto the perovskite layer. "Thus, we can obtain fairly pure doped organic materials for efficient hole transporting layers."

The team, which included researchers from Samsung, Yale University, Korea Research Institute of Chemical Technology, The Graduate Center of the City University, Wonkwang University, and the Gwangju Institute of Science and Technology also found that the CO₂ doping method can be used for p-type doping of other π -conjugated polymers, such as PTAA, MEH-PPV, P3HT, and PBDB-T. According to Taylor the researchers are looking to push the boundary beyond typical organic semiconductors used for solar cells.

"We believe that wide applicability of CO₂ doping to various π -conjugated organic molecules stimulates research ranging from organic solar cells to organic light emitting diodes (OLEDs) and organic field effect transistors (OFETs), even to thermoelectric devices that all

require controlled doping of organic semiconductors," Taylor explained, adding that since this process consumes quite a large amount of CO₂ gas, it can be also considered for CO₂ capture and sequestration studies in the future.

"At a time when governments and companies alike are now looking to reduce CO₂ emissions if not to de-carbonize, this research offers an avenue for reacting large amounts of CO₂ in lithium carbonate to improve next generation solar cells, while removing this greenhouse gas from the atmosphere," he explained, adding that the idea for this novel approach was a counterintuitive insight from the team's battery research.

"From our long history of working with lithium [oxygen/air batteries](#) we know that lithium carbonate formation from exposure of oxygen electrodes to air is a big challenge because it depletes the battery of lithium ions, which destroys battery capacity. In this spiro doping reaction, however, we are actually exploiting lithium carbonate formation, which binds [lithium](#) and prevents it from becoming mobile ions detrimental to the long term stability of the perovskite solar cell. We are hoping that this CO₂ doping technique could be a stepping stone for overcoming existing challenges in organic electronics and beyond."

More information: Kong, J. et al. CO₂ doping of organic interlayers for perovskite solar cells. *Nature* (2021). [DOI: 10.1038/s41586-021-03518-y](#)

Provided by NYU Tandon School of Engineering

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