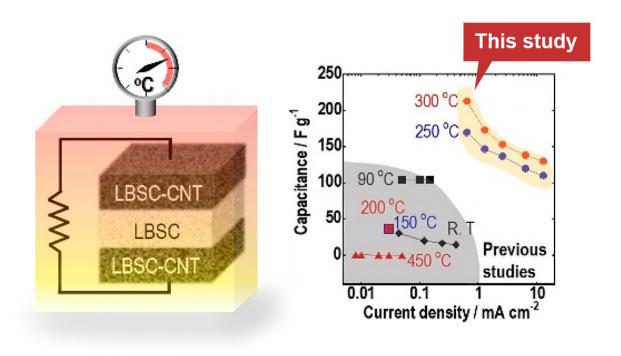


## New heat-tolerant, high-capacity capacitor created with solid electrolytes borrowed from all-solid-state batteries

August 19 2022



Bulk-type symmetric all-solid-state capacitor, with a LBSC SE layer between two electrode layers of an LBSC-CNT composite (left), showed low resistance and were operable at 100–300 °C (right). Credit: Hayashi, Osaka Metropolitan University



Capacitors are energy storage devices—consisting of two electrodes and an electrolyte—that are capable of rapid charging and discharging because of charge adsorption and desorption properties at the electrode-electrolyte interface. Because capacitors' energy storage does not involve chemical reactions, their storage capacity is lower than that of lithiumion batteries, but they are useful for power leveling for renewable energy that requires repeated charging at high currents, regenerative braking energy for trains and electric or hybrid cars, as well as instantaneous voltage drop compensation devices that prevent equipment failure due to lightning strikes. They are also expected to be used to store energy for wearable devices in the near future.

Most capacitors use a liquid electrolyte with a low boiling point, which can only be used at temperatures below 80°C. Ceramic capacitors that use solid inorganic materials as a dielectric can be used at temperatures above 80°C, but their storage capacity is much lower than liquid electrolyte capacitors, which limits their use to electronic circuits.

To increase the energy storage of capacitors, it is necessary to have a large contact area at the interface between the electrode and the electrolyte. Making a large contact area is difficult using solid electrolytes; so, the creation of a capacitor with high storage capacity that can also operate at high temperatures has been desired for a long time.

A research group led by Professor Akitoshi Hayashi at the Graduate School of Engineering, Osaka Metropolitan University, has developed a solid electrolyte that is highly deformable, allowing it to have a large contact area with an electrode, which was developed to be used for an oxide-based all-solid-state battery.

In this study, they fabricated a composite using the same highly deformable solid electrolyte and carbon, then used it to construct both



electrodes for a bulk-type all-solid-state capacitor. This capacitor is capable of high current densities and high-capacity charging and discharging at temperatures of 200-300°C, creating the world's first bulk-type all-solid-state capacitors. The researchers expect that their capacitor will be used to improve technology for high-temperature environments, which could not be developed previously due to these technical limitations.

"The key to realizing this capacitor was to take the solid oxide electrolytes that we have been developing for all-solid-state lithium batteries—which combine excellent deformability and lithium-ion conductivity—and apply them to capacitors," explained Professor Hayashi.

In the future, the researchers hope to construct all-solid-state hybrid capacitors with even higher energy densities, by controlling the chemical reaction between a solid <u>electrolyte</u> and carbon, then combining them with positive electrode materials used in <u>lithium-ion batteries</u>.

The research was published in the Journal of Power Sources.

**More information:** Takashi Hakari et al, Thermally stable bulk-type all-solid-state capacitor with a highly deformable oxide solid electrolyte, *Journal of Power Sources* (2022). <u>DOI:</u> 10.1016/j.jpowsour.2022.231821

## Provided by Osaka Metropolitan University

Citation: New heat-tolerant, high-capacity capacitor created with solid electrolytes borrowed from all-solid-state batteries (2022, August 19) retrieved 20 March 2024 from <a href="https://techxplore.com/news/2022-08-heat-tolerant-high-capacity-capacitor-solid-">https://techxplore.com/news/2022-08-heat-tolerant-high-capacity-capacitor-solid-</a>



## electrolytes.html

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.