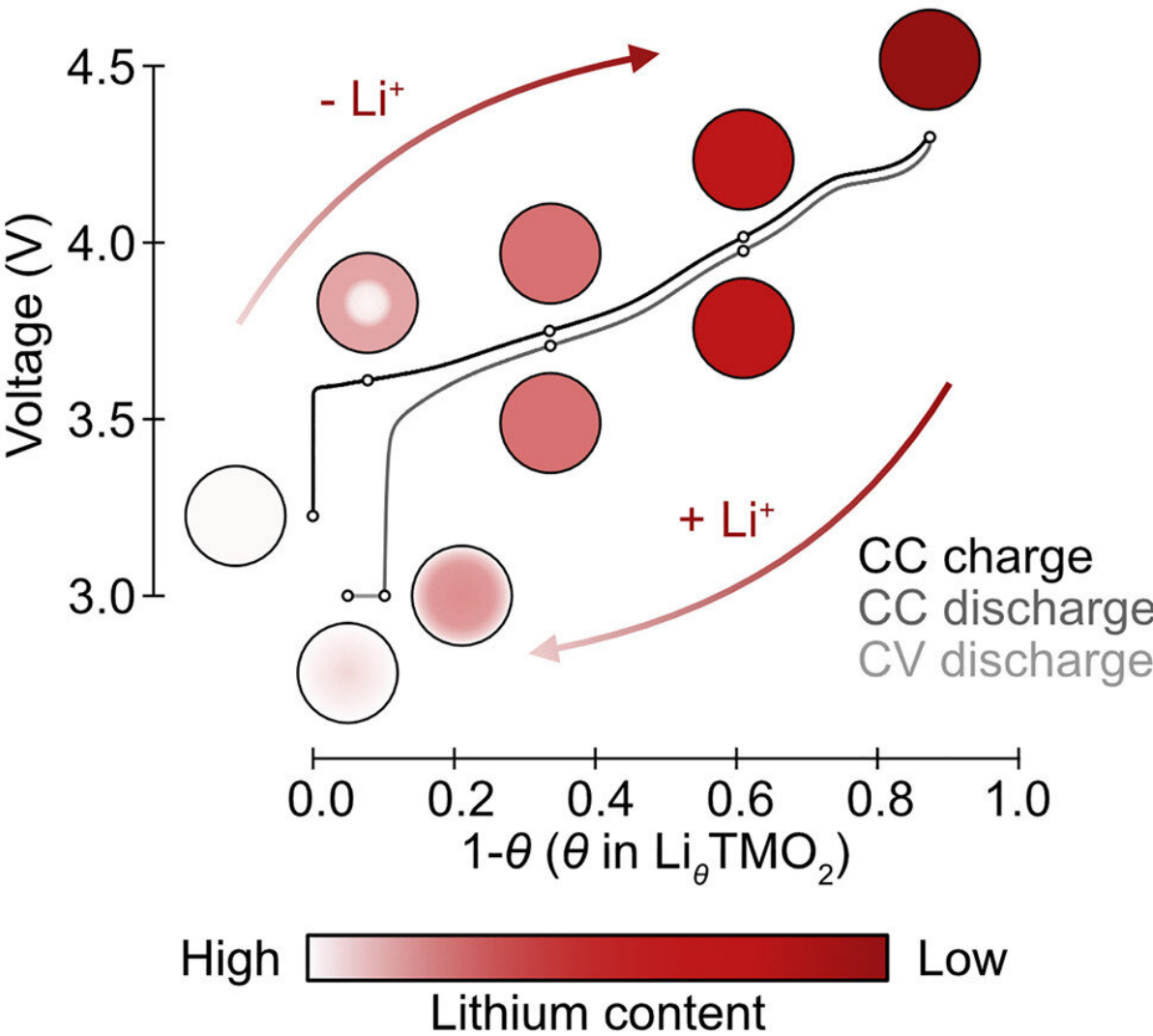


Watching lithium in real time could improve performance of EV battery materials

October 14 2022



Graphical abstract. Credit: *Joule* (2022). DOI: 10.1016/j.joule.2022.09.008

Researchers have found that the irregular movement of lithium ions in next-generation battery materials could be reducing their capacity and hindering their performance.

The team, led by the University of Cambridge, tracked the movement of lithium ions inside a promising new battery material in real time.

It had been assumed that the mechanism by which lithium ions are stored in battery materials is uniform across the individual active particles. However, the Cambridge-led team found that during the charge-discharge cycle, lithium storage is anything but uniform.

When the battery is near the end of its discharge cycle, the surfaces of the active particles become saturated by lithium while their cores are lithium deficient. This results in the loss of reusable lithium and a reduced capacity.

The research could help improve existing battery materials and could accelerate the development of next-generation batteries. The results are published in *Joule*.

Electrical vehicles (EVs) are vital in the transition to a zero-carbon economy. Most [electric vehicles](#) on the road today are powered by [lithium-ion batteries](#), due in part to their high energy density.

However, as EV use becomes more widespread, the push for longer ranges and faster charging times means that current [battery materials](#) need to be improved, and new materials need to be identified.

Some of the most promising of these materials are state-of-the-art positive electrode materials known as layered lithium nickel-rich oxides, which are widely used in premium EVs. However, their working mechanisms, particularly lithium-ion transport under practical operating conditions, and how this is linked to their electrochemical performance, are not fully understood, so we cannot yet obtain maximum performance from these materials.

By tracking how light interacts with active particles during battery operation under a microscope, the researchers observed distinct differences in lithium storage during the charge-discharge cycle in nickel-rich manganese cobalt oxide (NMC).

"This is the first time that this non-uniformity in lithium storage has been directly observed in individual particles," said co-first author Alice Merryweather, from Cambridge's Yusuf Hamied Department of Chemistry. "Real time techniques like ours are essential to capture this while the battery is cycling."

Combining the experimental observations with computer modeling, the researchers found that the non-uniformity originates from drastic changes to the rate of lithium-ion diffusion in NMC during the charge-discharge cycle. Specifically, lithium ions diffuse slowly in fully lithiated NMC particles, but the diffusion is significantly enhanced once some lithium ions are extracted from these particles.

"Our model provides insights into the range over which lithium-ion diffusion in NMC varies during the early stages of charging," said co-first author Dr. Shrinidhi Pandurangi from Cambridge's Department of Engineering. "Our model predicted lithium distributions accurately and captured the degree of heterogeneity observed in experiments. These predictions are key to understanding other battery degradation mechanisms such as particle fracture."

Importantly, the [lithium](#) heterogeneity seen at the end of discharge establishes one reason why nickel-rich cathode materials typically lose around ten percent of their capacity after the first charge-discharge cycle.

"This is significant, considering one industrial standard that is used to determine whether a [battery](#) should be retired or not is when it has lost 20 percent of its capacity," said co-first author Dr. Chao Xu, from ShanghaiTech University, who completed the research while based at Cambridge.

More information: Chao Xu et al, Operando visualization of kinetically induced lithium heterogeneities in single-particle layered Ni-rich cathodes, *Joule* (2022). [DOI: 10.1016/j.joule.2022.09.008](https://doi.org/10.1016/j.joule.2022.09.008)

Provided by University of Cambridge

Citation: Watching lithium in real time could improve performance of EV battery materials (2022, October 14) retrieved 23 April 2024 from <https://techxplore.com/news/2022-10-lithium-real-ev-battery-materials.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.