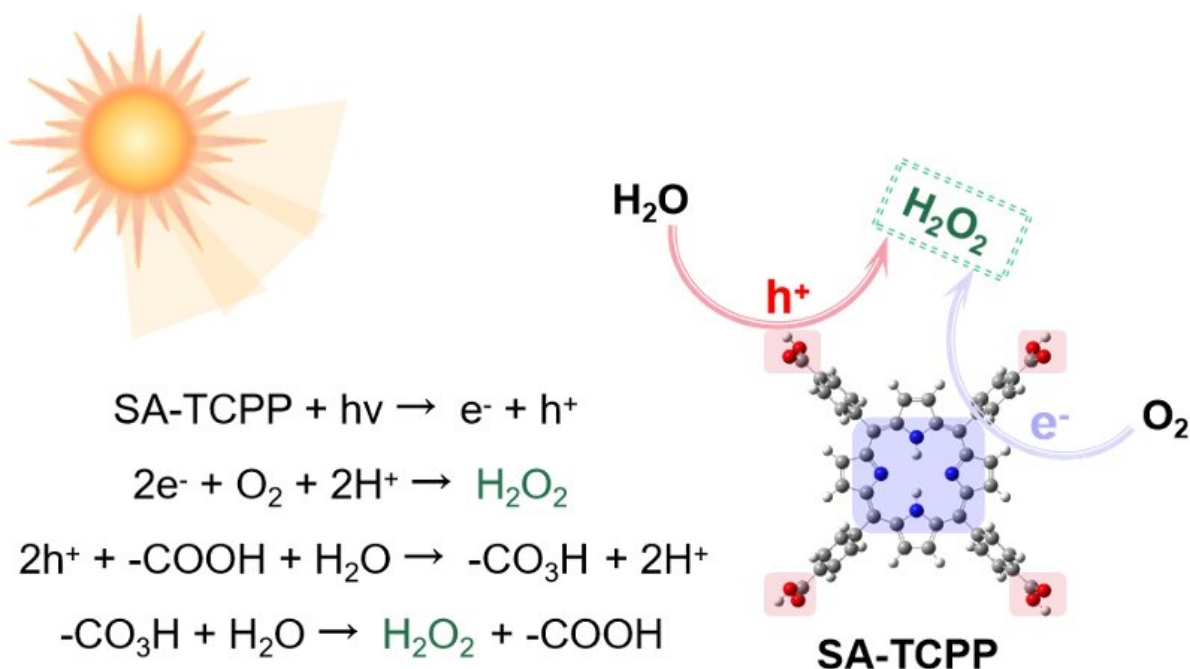


# A photocatalyst that can produce hydrogen peroxide from oxygen and water

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The proposed schematics for  $H_2O_2$  production on SA-TCPP supramolecular photocatalysts. Credit: Zhang et al

Hydrogen peroxide ( $H_2O_2$ ) is a colorless liquid compound with strong oxidizing properties that can have numerous industrial and medical applications. This compound is generally synthesized through the anthraquinone process, through which the compound anthraquinone combined with palladium-based materials catalyzes the oxidation of

dihydrogen ( $\text{H}_2$ ) to  $\text{H}_2\text{O}_2$  with  $\text{O}_2$  from air.

$\text{H}_2\text{O}_2$  could also theoretically be synthesized through photocatalysis, a chemical reaction that entails the absorption of light by reacting elements through the addition of catalysts. However, this has been difficult to realize reliably using existing catalysts, which may respond weakly to sunlight, require sacrificial reagents or are not sufficiently active during the reaction.

Researchers at Jiangnan University and Tsinghua University recently introduced a new [photocatalyst](#) that could be used to reliably produce  $\text{H}_2\text{O}_2$  from only  $\text{H}_2\text{O}$  and  $\text{O}_2$  through photocatalysis. This photocatalyst, presented in a paper in *Nature Energy*, is based on self-assembled tetrakis(4-carboxyphenyl)porphyrin, a compound sometimes used to analyze or remove metals.

"Photocatalytic  $\text{H}_2\text{O}_2$  production only requires  $\text{H}_2\text{O}$ ,  $\text{O}_2$  and sunlight, which is considered as a promising alternative to the industrial anthraquinone method," Chengsi Pan, one of the researchers who carried out the study, told Tech Xplore. "However, due to a high Gibbs [free energy](#) of  $117 \text{ kJ mol}^{-1}$ , most photocatalytic systems require sacrificial reagents (e.g., EtOH, IPA) to generate considerable  $\text{H}_2\text{O}_2$ . Our group has been working since 2018 to develop a photocatalyst that can continuously produce  $\text{H}_2\text{O}_2$  without sacrificial reagents."

In [one of their previous works](#), Pan and his colleagues had tested the potential of bismuth phosphate ( $\text{BiPO}_4$ ) as a photocatalyst for producing  $\text{H}_2\text{O}_2$ . While their results were promising, they found that  $\text{BiPO}_4$  only responded to UV light, which limited its use and prevented solar energy-based applications. In their new study, they thus set out to evaluate another photocatalyst, which was based on the narrow bandgap tetrakis(4-carboxyphenyl)porphyrin supramolecule (TCPP).

"We employed a self-assembled tetrakis(4-carboxyphenyl)porphyrin nanosheet (SA-TCPP) to achieve the desired activity," Pan explained. "This material was prepared by dissolving commercial bulk TCPP in NaOH and subsequently adding HCl. For the H<sub>2</sub>O<sub>2</sub> generation experiments, a certain amount of SA-TCPP was added to a glass bottle containing only water and O<sub>2</sub> bubbling."

To evaluate the activity of their photocatalyst, Pan and his colleagues irradiated the glass bottle with the solution at an air mass 1.5 spectra (AM 1.5G) using a solar simulator, and then heated it at an elevated temperatures of 353 K. In initial tests, the SA-TCPP photocatalyst exhibited a near-infrared wavelength quantum efficiency of 1.1% at 940 nm and a high solar-to-chemical conversion efficiency of ca. 1.2%.

"Encouraged by these results, we evaluated the photocatalysts in a lab-made flow reactor, which resulted in a H<sub>2</sub>O<sub>2</sub> accumulation concentration of ca. 1.1 wt%, a level close to the commercial one (3 wt%), which is the first example that produces a commercial level of H<sub>2</sub>O<sub>2</sub> via the photocatalytic method," Pan said.

"We then also discovered that SA-TCPP could perform H<sub>2</sub>O<sub>2</sub> generation through a dual pathway of O<sub>2</sub> reduction and H<sub>2</sub>O oxidation, a unique characteristic compared to most previously reported photocatalysts. Notably, we observed that the introduction of -COOH can help the H<sub>2</sub>O oxidation pathway by converting to -CO<sub>3</sub>H groups, which subsequently undergo thermal hydrolysis."

The results of initial tests conducted by this team of researchers are highly promising, highlighting the value of their proposed photocatalyst for the large-scale production of H<sub>2</sub>O<sub>2</sub> using solar light. In the future, their work could inform the design of other organic photocatalysts that can achieve even higher solar efficiencies, enabling the non-sacrificial generation of H<sub>2</sub>O<sub>2</sub>.

"We now plan to explore two potential areas for future work: the integration of our photocatalyst with photothermal conversion materials, and the redesign of the reactor to achieve increased production amount beyond bench scale," Pan added.

"By combining our photocatalyst with photothermal conversion materials, we can raise its local temperature only through light irradiation, reducing the need for external heat sources as reported in our current paper. Additionally, we aim to develop a scalable prototype device for H<sub>2</sub>O<sub>2</sub> production with SA-TCPP using only air, H<sub>2</sub>O, and sunlight by optimizing the reactor design and reaction parameters."

**More information:** Yaning Zhang et al, H<sub>2</sub>O<sub>2</sub> generation from O<sub>2</sub> and H<sub>2</sub>O on a near-infrared absorbing porphyrin supramolecular photocatalyst, *Nature Energy* (2023). [DOI: 10.1038/s41560-023-01218-7](https://doi.org/10.1038/s41560-023-01218-7)

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