

## **Researchers make progress towards highperformance, aqueous zinc-ion batteries**

May 8 2023, by Liu Jia



Credit: Pixabay/CC0 Public Domain

A team led by Prof. Song Li from the University of Science and Technology of China (USTC) of the Chinese Academy of Sciences (CAS) proposed the concept of intercalant-induced V  $t_{2g}$  orbital occupation and developed an NH<sub>4</sub><sup>+</sup>-intercalated vanadium oxide



 $(NH_4^+-V_2O_5)$  for high-performance aqueous zinc-ion batteries (ZIBs). This work was published in *PNAS*.

ZIBs, due to their safety, non-toxicity, and high theoretical capacity, have become one of the most promising sustainable energy storage technologies. Among various electrode materials for ZIBs, vanadium oxides have been widely studied as cathode materials for ZIBs due to their flexible crystal structure and multivalence of vanadium.

Based on the ion or molecular pre-intercalation strategy, the problems of insufficient lattice space and low electronic conductivity of cathode materials can be effectively solved, thereby further improving battery performance.

However, current research on intercalation cathode materials mostly focuses on the enhancement of capacity by interlayer space expansion. Therefore, it is important to develop advanced in-situ characterization techniques, which assist in studying the intrinsic structural variation of electrode materials induced by intercalants from the perspective of atomic orbitals This will become the key to the design of highperformance cathode materials in the future.

In this work, researchers introduced a variety of in-situ and ex-situ synchrotron radiation spectroscopy techniques, revealing the change of V  $3dt_{2g}$  orbital occupation in V<sub>2</sub>O<sub>5</sub> after the intercalation of NH<sub>4</sub><sup>+</sup>, as well as the reversible evolution law during the charge and discharge process.

They found that  $NH_4^+$  intercalation largely induced the structural distortion of the V-O bond, which further led to a rearrangement of the electronic structure and facilitated the occupation of the 3dxy vacancy state in the  $Vt_{2g}$  orbital. The  $Vt_{2g}$  orbital occupation greatly improved the electrical conductivity of the material.



Additionally, the researchers found that  $NH_4^+$  intercalation resulted in the broadened layer spacing, which significantly accelerated the <u>electron</u> <u>transfer</u> and Zn-ion migration, realizing the ultra-high multiplicity performance of the zinc ion battery.

The experimental results showed that the specific capacity of the ammonium intercalated vanadium pentoxide  $(NH_4^+-V_2O_5)$  cathode material remains the rate capability at 101.0 mA h g-1 at a current density of 200 C, with a charging time of 18 s.

This study provides a better understanding of the mechanism of storing  $Zn_2^+$  in intercalated  $V_2O_5$  materials, and lays a foundation for the design of high-performance intercalated <u>cathode materials</u> of ZIBs.

**More information:** Yixiu Wang et al, Intercalant-induced V t 2 g orbital occupation in vanadium oxide cathode toward fast-charging aqueous zinc-ion batteries, *Proceedings of the National Academy of Sciences* (2023). DOI: 10.1073/pnas.2217208120

Provided by Chinese Academy of Sciences

Citation: Researchers make progress towards high-performance, aqueous zinc-ion batteries (2023, May 8) retrieved 6 May 2024 from <u>https://techxplore.com/news/2023-05-high-performance-aqueous-zinc-ion-batteries.html</u>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.