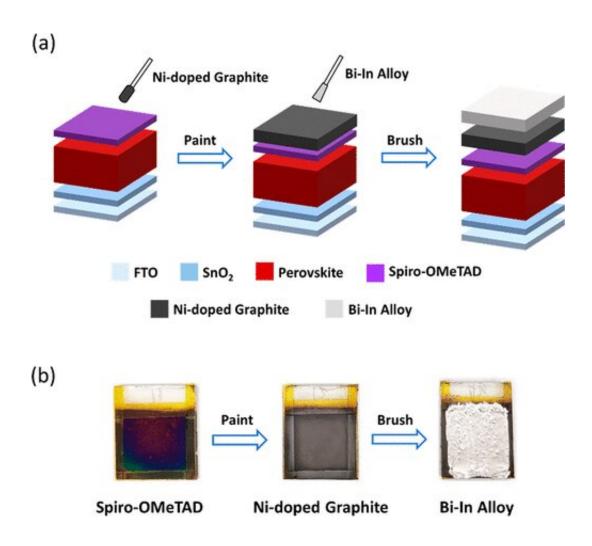


Substitute for gold layer in perovskite clears way for cheaper commercialization

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(a) Fabrication of a graphite-alloy bilayer back electrode on a PSC device. (b) Top-view image of spiro-OMeTAD, Ni-doped graphite, and Bi-In alloy. Credit: *ACS Energy Letters* (2023). DOI: 10.1021/acsenergylett.3c00852



A thin layer of gold or silver helps improve the efficiency of perovskite solar cells, but researchers have found a less expensive material that will enable commercialization of the technology without exorbitant cost.

"A layer of gold in a solar panel or even a layer of silver is probably too expensive," said Kai Zhu, a senior scientist in the Chemistry and Nanoscience Center at the U.S. Department of Energy's National Renewable Energy Laboratory (NREL). "It would make the solar panel not affordable for most people."

Instead of a precious metal, Zhu and his colleague at Northern Illinois University have a better, cheaper idea. The solution is a nickel-doped graphite layer coupled with a bismuth-indium alloy layer. The two layers can be easily integrated into the <u>perovskite</u> device through painting them on, offering a low-cost fabrication method.

The findings are detailed in a new paper, "Nickel-Doped Graphite and Fusible Alloy Bilayer Back Electrode for Vacuum-Free Perovskite Solar Cells," published in the journal *ACS Energy Letters*.

"Our team has identified a potentially disruptive technology that could help reduce the <u>infrastructure investment</u> for use of highly promising perovskite <u>solar cells</u> in <u>solar panels</u>," said Tao Xu, a chemistry professor at Northern Illinois University and Zhu's co-corresponding author of the new paper.

"Our approach replaces costly gold, commonly used to make the backmetal electrode in these solar cells through an expensive hightemperature vacuum-chamber process. Instead of gold, we use inexpensive materials that can be readily laminated to thin films at atmospheric pressure and mild temperatures. We think this will be an appealing low-cost solution that could help speed commercialization of <u>perovskite solar cells</u>."



Xu and Zhu have previously collaborated on ways to sequester lead should a perovskite solar cell become damaged, as the cells contain a minute amount of the element. In addition to Zhu's NREL colleague So Yeon Park, the other co-authors are elsewhere in Illinois, at Northwestern University and Argonne National Laboratory.

Zhu said the perovskite solar cell made with the new materials yielded a laboratory efficiency of 21%. Further research should boost the efficiency higher and closer to that of perovskites made with <u>precious</u> <u>metals</u>, which hold a record efficiency of 26%. Metal is better than carbon at conducting electricity.

Perovskite solar <u>cells</u> are made by depositing chemicals onto a substrate. Individual layers all serve a purpose, with the perovskite layer serving as the semiconductor. The photons from sunlight trigger electrons to move in one direction, creating a vacancy—or "hole"—that then moves in the opposite direction. The right energy level is needed to spark this movement, in the process creating an electric current. The graphite material and gold both possess the right energy level.

By eliminating a layer of precious metals, the cost to manufacturer perovskites will be significantly cheaper, the researchers calculated. An analysis suggests that for a perovskite-based solar plant with a gigawatt of power output, using the graphite/alloy bilayer will slash costs of the contact electrodes by a factor of between 4 and 1,000 depending on the other types of materials used for back materials.

"That's the selling point for this approach," Zhu said.

More information: Mengyuan Li et al, Nickel-Doped Graphite and Fusible Alloy Bilayer Back Electrode for Vacuum-Free Perovskite Solar Cells, *ACS Energy Letters* (2023). DOI: 10.1021/acsenergylett.3c00852



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