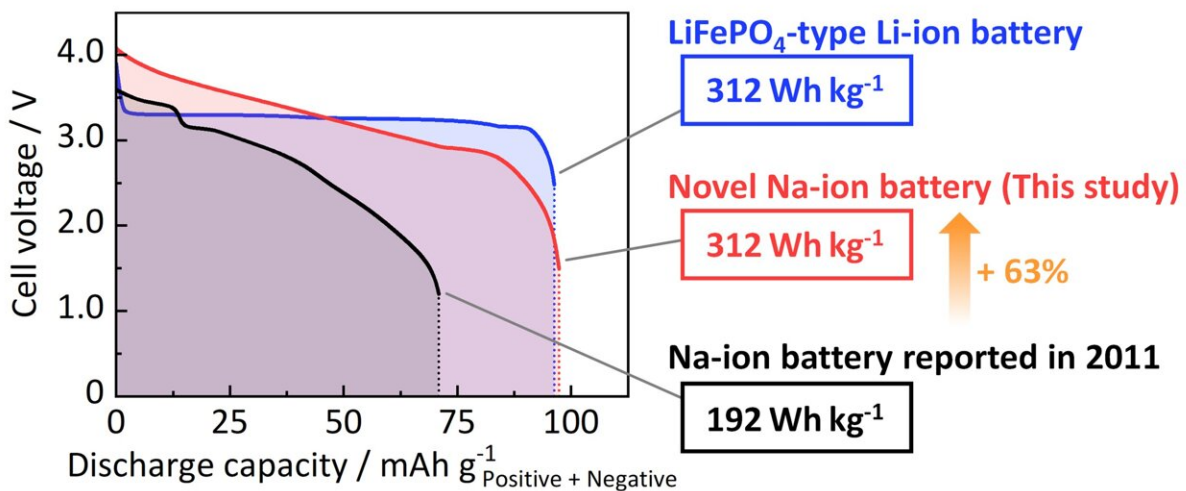


# Template for success: Shaping hard carbon electrodes for next-generation batteries

November 13 2023



It is possible to incorporate nanopores in hard carbon by using zinc oxide as a template during its synthesis. These pores enable the material to store many more charge carriers, making it a promising electrode candidate for sodium-ion batteries that can reach an energy density comparable to that of LiFePO<sub>4</sub>-type lithium-ion batteries. Credit: Shinichi Komaba, Tokyo University of Science (TUS), Japan

Lithium-ion batteries (LIBs) are, by far, the most widely used type of rechargeable batteries, spanning numerous applications. These include consumer electronics, electric vehicles, renewable energy systems, and spacecraft.

Although LIBs deliver the best performance in many aspects when compared to other rechargeable batteries, they have their fair share of disadvantages. Lithium is a rather scarce resource, and its price will rise quickly with its availability dwindling in the future. Moreover, lithium extraction and improperly discarded LIBs pose huge environmental challenges as the liquid electrolytes commonly used in them are toxic and flammable.

The shortcomings of LIBs have motivated researchers worldwide to look for alternative energy storage technologies. Sodium (Na)-ion batteries (NIBs) and potassium-ion batteries (KIBs) are two rapidly emerging options that are cost-efficient as well as sustainable. Both NIBs and KIBs are projected to be billion-dollar industries by the end of the decade.

Governments around the world, including those of the US, Austria, Hong Kong, Germany, and Australia, are promoting research and innovation in this field. Moreover, companies such as Faradion Limited, TIAMAT SAS, and HiNa Battery Technology Co. Ltd., are investing heavily in this technology. Both Contemporary Amperex Technology Co. Limited and Build Your Dreams are expected to introduce electric vehicle battery packs with NIBs soon.

Unfortunately, however, the capacity of the electrode materials used in NIBs and KIBs still lags behind that of LIBs. Against this backdrop, a research team led by Professor Shinichi Komaba from Tokyo University Science (TUS), Japan, has been working to develop groundbreaking high-capacity electrode materials for NIBs and KIBs.

In their latest study, [published in \*Advanced Energy Materials\*](#), they report a new synthesis strategy for nanostructured "hard carbon" (HC) electrodes that deliver unprecedented performance. The study was co-authored by Mr. Daisuke Igarashi, Ms. Yoko Tanaka, and Junior Associate Professor Ryoichi Tatara from TUS, and Dr. Kei Kubota from

the National Institute for Materials Science (NIMS), Japan.

But what is HC and why is it useful for NIBs and KIBs? Unlike other forms of carbon, such as graphene or diamond, HC is amorphous; it lacks a well-defined crystalline structure. Additionally, it is strong and resistant.

In an earlier 2021 study, Prof. Komaba and his colleagues found a way to use magnesium oxide (MgO) as a template during the synthesis of HC electrodes for NIBs, altering their final nanostructure. The process had led to the formation of nanopores within the electrodes upon MgO removal, which, in turn, had vastly increased their capacity to store  $\text{Na}^+$  ions.

Motivated by their previous findings, the researchers explored whether compounds made from zinc (Zn) and calcium (Ca) could also be useful as nano-templates for HC electrodes. To this end, they systematically investigated different HC samples made using zinc oxide (ZnO) and calcium carbonate ( $\text{CaCO}_3$ ) and compared their performance with the ones synthesized using [magnesium oxide](#) (MgO).

Preliminary experiments showed that ZnO was particularly promising for the negative electrode of NIBs. Accordingly, the researchers optimized the concentration of ZnO embedded in the HC matrix during synthesis, demonstrating a reversible capacity of  $464 \text{ mAh g}^{-1}$  (corresponding to  $\text{NaC}_{4.8}$ ) with a high initial Coulombic efficiency of 91.7% and a low average potential of 0.18 V vs.  $\text{Na}^+/\text{Na}$ .

The team achieved remarkable results by incorporating this powerful electrode material into an actual battery.

"The NIB fabricated using the optimized ZnO-templated HC as the negative [electrode](#) exhibited an energy density of  $312 \text{ Wh kg}^{-1}$ ," says

Prof. Komaba. "This value is equivalent to the energy density of certain types of currently commercialized LIBs with  $\text{LiFePO}_4$  and graphite and is more than 1.6 times the [energy density](#) of the first NIBs ( $192 \text{ Wh kg}^{-1}$ ), which our laboratory reported back in 2011."

Notably, the ZnO-templated HC also exhibited a significant capacity of  $381 \text{ mAh g}^{-1}$  when incorporated into a KIB, further showcasing its potential.

Taken together, the results of this study show that using inorganic nanoparticles as a template to control the pore structure may provide an effective guideline for the development of HC electrodes. "Our findings prove that HCs are promising candidates for negative electrodes as an alternative to graphite," concludes Prof. Komaba.

In turn, this could make NIBs viable for practical applications, such as the development of sustainable [consumer electronics](#) and [electric vehicles](#) as well as low carbon footprint energy storage systems for storing energy from solar and wind farms.

**More information:** Daisuke Igarashi et al, New Template Synthesis of Anomalously Large Capacity Hard Carbon for Na- and K-Ion Batteries, *Advanced Energy Materials* (2023). [DOI: 10.1002/aenm.202302647](https://doi.org/10.1002/aenm.202302647)

Provided by Tokyo University of Science

Citation: Template for success: Shaping hard carbon electrodes for next-generation batteries (2023, November 13) retrieved 28 April 2024 from <https://techxplore.com/news/2023-11-template-success-hard-carbon-electrodes.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private

study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.