

Converting a steel mill to climate-neutral steel production

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Direct reduction plant on a laboratory scale for reaction kinetic investigations and shaft furnace modeling. Credit: Fraunhofer-Gesellschaft

Steel production causes significant emissions of carbon dioxide. To decarbonize steel production and its high carbon dioxide emissions, Fraunhofer researchers, TS ELINO GmbH and Salzgitter AG are working on converting an existing steel mill to climate-neutral production methods.

The aim is to produce [steel](#) by the direct reduction of [iron ore](#) with hydrogen, which would completely replace harmful coke as a reducing agent. The hydrogen required for this method is produced using electrolysis processes with electricity generated from renewable energy sources. Overall, this could reduce [carbon dioxide emissions](#) by up to 97%. The decarbonization of the steel industry would therefore contribute significantly to protecting the climate.

Steel has many useful qualities and is found in virtually all products where strength, malleability and stability are key properties. Whether in buildings, vehicles, machines or [household appliances](#), steel is an indispensable component—but according to the Competence Center on Climate Change Mitigation in Energy-Intensive Industries (KEI), about 55 million tons of CO₂ are emitted every year from steel production in Germany alone. The steel industry is therefore responsible for around 28% of the total CO₂ emissions generated by German industry. This is primarily due to the use of coke, which is required in blast furnaces to remove oxygen from the iron ore and to extract the pig iron.

Work has been underway to develop new technologies to decarbonize production for years. The Fraunhofer Institute for Ceramic Technologies

and Systems IKTS, the partner institutes Fraunhofer Institute for Systems and Innovation Research ISI and Fraunhofer Institute for Environmental, Safety and Energy Technology UMSICHT as well as Salzgitter AG are focusing on hydrogen-based direct reduction.

During direct reduction in a reactor, the iron ore reacts with hydrogen at a high temperature. This hydrogen acts as a reducing agent and removes the [iron oxide](#) from the iron ore. What remains is the raw pig iron. "By using electricity from renewable energy sources, the production of hydrogen is completely free of CO₂ emissions. This means that by implementing green hydrogen in crude steel production, we can save up to 97% of harmful carbon dioxide," explains Dr. Matthias Jahn, head of department for Energy and Process Engineering at Fraunhofer IKTS.



µDRAL plant—first direct reduction plant flexibly operated with hydrogen and

natural gas in an integrated mill. Credit: Fraunhofer-Gesellschaft

Green hydrogen from high-temperature electrolysis

In principle, green hydrogen is produced using electrolysis, where an electrical voltage is applied to split steam into hydrogen and oxygen. In particular, high-temperature electrolysis based on solid oxide electrolysis cells (SOEC) has had a long history at Fraunhofer IKTS. Given that the waste heat from high-temperature processes can be used to increase the electrical efficiency, this offers considerable advantages compared to other electrolysis processes—especially in steel production.

Fraunhofer IKTS researchers have developed in-house electrolysis cells and stacks and used their own operating data for the techno-economic assessment of the process concept. Furthermore, the consortium has demonstrated the feasibility of the new process in both the MACOR study and BeWiSe follow-up project.

Demonstration plant on the company premises

Having launched their new BeWiSe project, the researchers and their industrial partners are now further optimizing the entire process chain in terms of resource and energy efficiency. For this purpose, a direct reduction demonstration plant measuring approx. 30 meters in height on the Salzgitter AG premises is also used.

For example, the project is investigating how biogenic materials can be used as a substitute for coal and [natural gas](#) to adjust the carbon content required in steel. Considering that large quantities of water are required, among other things, for electrolysis during the production of green steel, the project is also focusing on how to optimize the use of water. As a

result, the water formed during iron ore reduction with hydrogen is to be processed as far as possible for reuse.

The researchers are contributing all of their expertise, which ranges from high-temperature electrolysis, membrane processes for gas separation and water treatment to process simulation and plant modeling using digital twins. "We are not satisfied with simply providing selective solutions and laboratory plants. We aim to offer Salzgitter AG comprehensive technological support for all process steps," states Gregor Herz, group manager for Modeling and Simulation at Fraunhofer IKTS.

Dr. Alexander Redenius, head of Resource Efficiency and Technology Development at Salzgitter Mannesmann Forschung adds, "We have been working successfully with the Fraunhofer researchers for six years to transform steel production. The direct reduction demonstration plant enables us to optimize the reduction process and how it interacts with the other process steps. Through this work, we are creating the basis for sustain-able steel production with low CO₂ emissions." The company aims to convert a third of its [steel production](#) to the climate-friendly process with [hydrogen](#) as early as 2026.

Provided by Fraunhofer-Gesellschaft

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