

## Synthesis process boosts perovskite solar cell performance to near market-ready standards

June 13 2024



2D perovskite sample. Credit: Jeff Fitlow/Rice University



Solar power is not only the <u>fastest growing</u> energy technology in recent history but also one of the <u>cheapest energy</u> sources and the <u>most</u> <u>impactful</u> in terms of reducing greenhouse gas emissions.

A Rice University <u>study</u> featured on the cover of *Science* describes a way to synthesize formamidinium lead iodide (FAPbI<sub>3</sub>)—the type of crystal currently used to make the highest-efficiency <u>perovskite solar cells</u> —into ultrastable, high-quality photovoltaic films. The overall efficiency of the resulting FAPbI<sub>3</sub>solar cells decreased by less than 3% over more than 1,000 hours of operation at temperatures of 85 degrees Celsius (185 Fahrenheit).

"Right now, we think that this is state of the art in terms of stability," said Rice engineer Aditya Mohite, whose lab has achieved progressive improvements in the perovskites' durability and performance over the past several years. "Perovskite solar cells have the potential to revolutionize <u>energy production</u>, but achieving long-duration stability has been a significant challenge."

With this most recent breakthrough, Mohite and collaborators have reached a critical milestone toward making perovskite photovoltaics market-ready. The key was "seasoning" the FAPbI<sub>3</sub>precursor solution with a sprinkling of specially designed two-dimensional (2D) perovskites. These served as a template guiding the growth of the bulk/3D perovskite, providing added compression and stability to the crystal lattice structure.

"Perovskite crystals get broken in two ways: chemically—destroying the molecules that make up the crystal—and structurally—reordering the molecules to form a different crystal," said Isaac Metcalf, a Rice materials science and nanoengineering graduate student and a lead author



on the study.



2D perovskite sample. Credit: Jeff Fitlow/Rice University

"Of the various crystals that we use in solar cells, the most chemically stable are also the least structurally stable and vice versa. FAPbI <sub>3</sub> is on the structurally unstable end of that spectrum."

While more stable than  $FAPbI_3$  both chemically and structurally, 2D perovskites are typically not great at harvesting light, making them a poor choice of material for solar cells.



However, the researchers hypothesized that 2D perovskites used as templates for growing FAPbI <sub>3</sub> films might impart their stability to the latter. To test this idea, they developed four different types of 2D perovskites—two with a <u>surface structure</u> nearly indistinguishable from that of FAPbI <sub>3</sub> and two less well-matched—and used them to make different FAPbI<sub>3</sub>film formulations.

"The addition of well-matched 2D crystals made it easier for FAPbI<sub>3</sub> crystals to form, while poorly matched 2D crystals actually made it harder to form, validating our hypothesis," Metcalf said.

"FAPbI<sub>3</sub> films templated with 2D crystals were higher quality, showing less internal disorder and exhibiting a stronger response to illumination, which translated as higher efficiency."





Isaac Metcalf is a materials science and nanoengineering graduate student at Rice University and a lead author on a study featured on the cover of Science. Credit: Jeff Fitlow/Rice University

The 2D crystal templates improved not only the efficiency of FAPbI <sub>3</sub> solar cells but also their durability. While solar cells without any 2D crystals degraded significantly after two days of generating electricity from sunlight in air, solar cells with 2D templates did not start degrading even after 20 days. By adding an encapsulation layer to the 2D-templated solar cells, stability was further improved to timescales approaching commercial relevance.

These findings could have a transformative impact on light-harvesting, or photovoltaic, technologies, further reducing manufacturing costs and enabling the construction of solar panels with a simplified structure that are <u>lighter weight and more flexible</u> than their silicon-based counterparts.

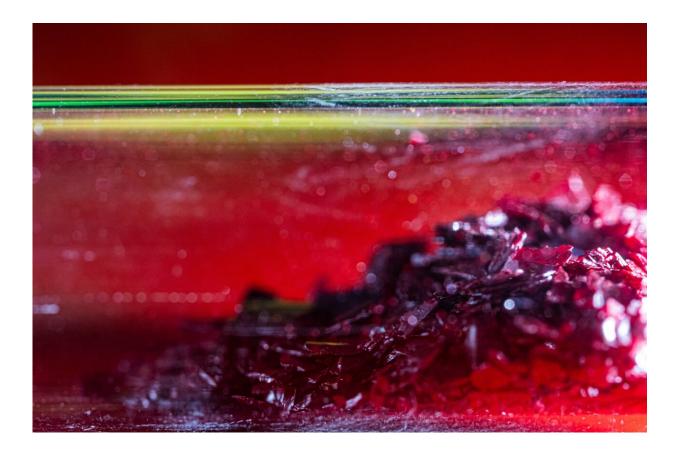
"Perovskites are soluble in solution, so you can take an ink of a perovskite precursor and spread it across a piece of glass, then heat it up and you have the absorber layer for a solar cell," Metcalf said.

"Since you don't need very high temperatures—perovskite films can be processed at temperatures below 150 Celsius (302 Fahrenheit)—in theory, that also means perovskite solar panels can be made on plastic or even flexible substrates, which could further reduce costs."

Although it is the most widely used semiconductor in photovoltaic cells, silicon entails manufacturing processes that are more resource-intensive than those of emerging alternatives. Among these, halide perovskites



stand out for their soaring efficiencies, which have gone from 3.9% in 2009 to over <u>26% currently</u>.



"Seasoning" the FAPbI<sub>3</sub> precursor solution with a sprinkling of specially designed two-dimensional (2D) perovskites improved not only the efficiency of FAPbI<sub>3</sub> solar cells but also their durability. Credit: Jeff Fitlow/Rice University

"It should be much cheaper and less energy-intensive to make highquality perovskite solar panels compared to high-quality silicon panels, because the processing is so much easier," Metcalf said.

"We need to urgently transition our global energy system to an emissionsfree alternative," he added, pointing to United Nations'



Intergovernmental Panel on Climate Change <u>estimates that</u> "make a strong case for solar as an alternative to fossil fuels."

Mohite underscored that advancements in solar energy technologies and infrastructure are critical for achieving the <u>greenhouse gas emissions</u> 2030 target and preventing a 1.5 degrees Celsius rise in global temperatures, which "would then set us on the right course to achieve <u>net zero carbon</u> emissions by 2050."

"If solar electricity doesn't happen, none of the other processes that rely on green electrons from the grid, such as thermochemical or electrochemical processes for chemical manufacturing, will happen," Mohite said. "Photovoltaics are absolutely critical."

Mohite is Rice's William M. Rice Trustee Professor, a professor of chemical and biomolecular engineering, and faculty director of the Rice Engineering Initiative for Energy Transition and Sustainability. In addition to Metcalf, Siraj Sidhik, a Rice doctoral alumnus, is a lead author on the study.

"I would like to give a lot of credit to Siraj, who started this project based on a theoretical idea by Professor Jacky Even at the University of Rennes," Mohite said. "I would also like to thank our collaborators at the national labs and at several universities in the U.S. and abroad whose help was instrumental to this work."

**More information:** Siraj Sidhik et al, Two-dimensional perovskite templates for durable and efficient formamidinium perovskite solar cells, *Science* (2024). DOI: 10.1126/science.abq6993. www.science.org/doi/10.1126/science.abq6993



## Provided by Rice University

Citation: Synthesis process boosts perovskite solar cell performance to near market-ready standards (2024, June 13) retrieved 18 June 2024 from https://techxplore.com/news/2024-06-synthesis-boosts-perovskite-solar-cell.html

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