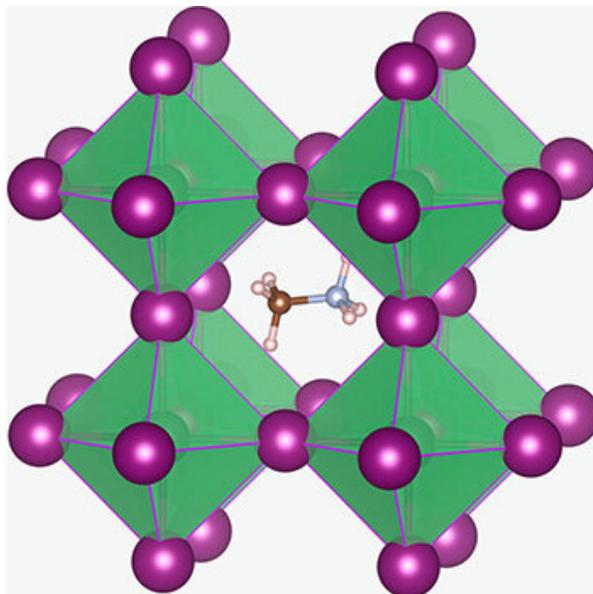


# A cool approach to defect-free solar cells

June 1 2020

---



The crystal structure of methylammonium lead halide, the material used to make the solar cell. Credit: Eames, C., Frost, J., Barnes, P. et al. Ionic transport in hybrid lead iodide perovskite solar cells. *Nature Communications* 6, 7497 (2015)

Chemists regularly use high temperatures to make things happen, but sometimes keeping cool can work better. Researchers at KAUST are discovering that a low-temperature approach can make better single crystals for use in solar cells.

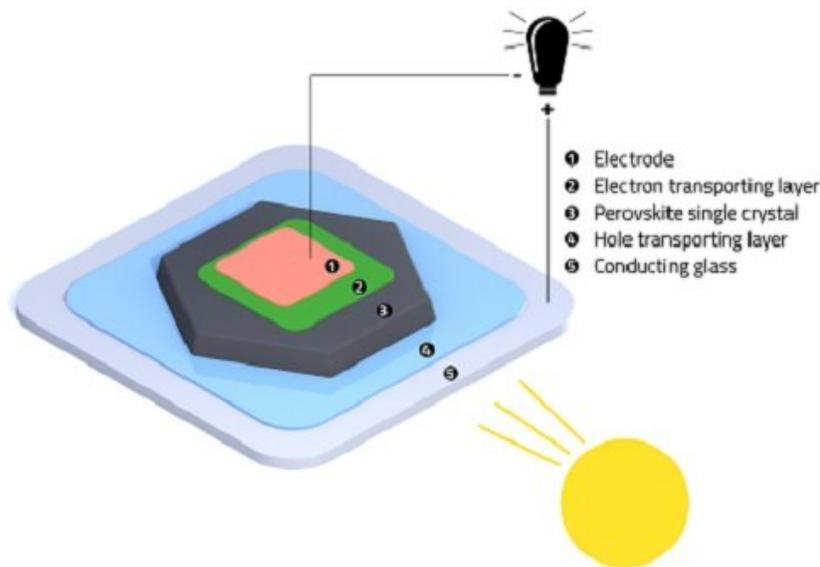
Osman M. Bakr's team in the KAUST Catalysis Center are finding many innovations in perovskites—materials that show great potential for

building new generations of solar [cells](#). Novel perovskites have positive and [negative ions](#) in the same arrangement as the natural [perovskite](#) calcium titanate ( $\text{CaTiO}_3$ ). Lead halide perovskites, containing lead ions and halide ions, such as chlorine and iodine in the perovskite mix, are attracting great interest for optoelectronic applications.

"Halide perovskite solar cells are considered the fastest growing photovoltaic technology," says Abdullah Alsalloum, masters student and co-first author of the study.

The crystals have previously been generated at high temperatures, however these have produced troublesome defects. Now, the KAUST team has devised a solution, in both senses of the word, allowing better crystals to form.

Broadly, their "solvent engineering" approach modifies the liquid solvent in which the components of the perovskite are initially dissolved, and in which they combine to build the perovskite crystals.



The structure of a perovskite solar cell that was created with a low temperature method. Credit: Abdullah Alsalloum

The specific perovskite they are developing is methylammonium lead iodide, containing large methylammonium ions ( $\text{CH}_3\text{NH}_3^+$ ), together with lead ( $\text{Pb}_2^+$ ) and iodide ( $\text{I}^-$ ) ions. Previously, the solutions used to make the crystals needed to be maintained at 120 degrees Celsius, but such a high temperature interfered with the optimal incorporation of methylammonium and iodide ions into the structure.

"We investigated a variety of different solvents to carry the perovskite components and allow them to crystallize," explains the other co-first author of the study Bekir Turedi. "Eventually, we found a solution that could work at less than 90 degrees Celsius and produce [single crystals](#) that were of significantly higher quality."

The enhancement in crystal formation was matched by improved performance when the crystals were used as the light-absorbing layers in solar cells.

"We obtained markedly higher photovoltages, and power conversion efficiencies approaching 22 percent, among the highest ever reported for these perovskites," says Bakr.

The researchers acknowledge that their single crystals are currently too small to be adopted for commercial use—a problem that currently hinders many other attempts to develop the full potential of perovskites in [solar cells](#). Nevertheless, this work at KAUST takes an important step toward tackling this issue of size as well as on further improvements.

"The field is still in its infancy and the most exciting parts are yet to

come," Alsalloum concludes.

**More information:** Abdullah Y. Alsalloum et al. Low-Temperature Crystallization Enables 21.9% Efficient Single-Crystal MAPbI<sub>3</sub> Inverted Perovskite Solar Cells, *ACS Energy Letters* (2020). [DOI: 10.1021/acseenergylett.9b02787](https://doi.org/10.1021/acseenergylett.9b02787)

Provided by King Abdullah University of Science and Technology

Citation: A cool approach to defect-free solar cells (2020, June 1) retrieved 26 April 2024 from <https://techxplore.com/news/2020-06-cool-approach-defect-free-solar-cells.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.